

Floating and sinking, Principle of floatation, application of floatation.

CLASS-VIIIC(RANK UP)

SUBJECT : PHYSICS

CHAPTER NUMBER: 2

CHAPTER NAME : PHYSICAL QUANTITIES AND MEASUREMENT

CHANGING YOUR TOMORROW

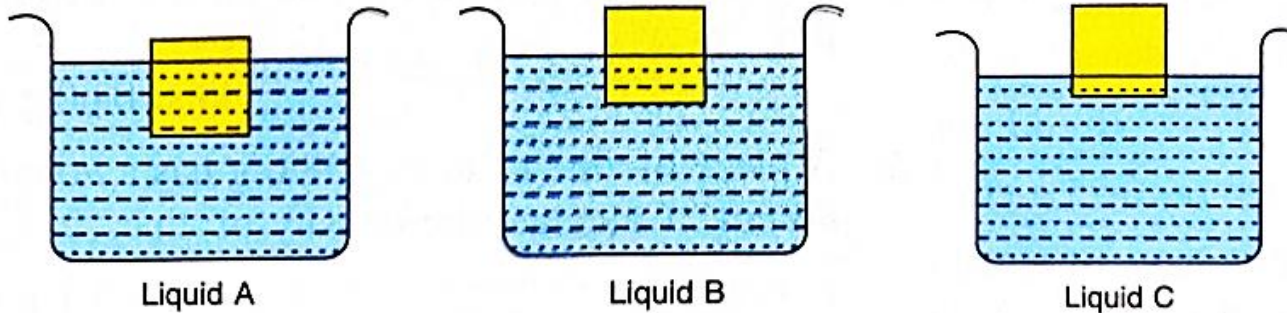
NUMERICAL

1. The mass of 10 cm^3 of silver is 103 gm. Find
 - (a) The density of silver in kg/m^3
 - (b) Relative density of silver.
2. A piece of wood of mass 150 g has a volume of 200 Cm^3 . Find the density of wood in C.GS. unit and S.I. unit.
3. How does the density of a liquid (or gas) vary with temperature?
4. Define the term relative density of a substance. What is the unit of relative density?
5. How does the density of a body and that of a liquid determine whether the body will float or sink into that liquid?
6. What is the law of floatation?

Numerical

Question: The diagram given below shows a body floating in three different liquids A, B and C at different levels.

- (a) In which liquid does the body experience the greatest buoyant force ?
- (b) Which liquid has the least density ?
- (c) Which liquid has the highest density ?



THANKING YOU
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