

Class	IX	Subject	CHEMISTRY
PD	7	Chapter-1	MATTER IN OUR SURROUNDING
Recapitulation of the previous class taught.	<ul style="list-style-type: none"> <li>❖ Scales of Temperature.</li> <li>❖ Plasma</li> <li>❖ Bose-Einstein Condensate (BEC)</li> </ul>		
Sub-Concepts	<ul style="list-style-type: none"> <li>▪ Evaporation and its factors.</li> <li>▪ Evaporation causes cooling.</li> <li>▪ Difference between evaporation and boiling.</li> </ul>		
Teaching Aid To be used	Smart Class, PowerPoint presentation, <b>classroom objects, (advertisements), charts.</b>		
Learning Outcome	<ul style="list-style-type: none"> <li>• Students will be able to know concept of evaporation.</li> <li>• They would know of the various factors affecting the rate of evaporation.</li> <li>• Students will be able to know about the difference between evaporation and boiling.</li> </ul>		
Sl. No	Step Wise (What to be done)		
1 Introduction	<p>For Achievers</p> <p>Teacher should initiate the discussion on following topics, which will revolve around the core topic of the chapter like, What's your view on Melting Point?</p> <ul style="list-style-type: none"> <li>➤ Vision to acquire knowledge of the evaporation and its factors.</li> <li>➤ They would get know how evaporation causes cooling.</li> <li>➤ They acquire knowledge</li> </ul>	<p>For Average</p> <ul style="list-style-type: none"> <li>➤ They would made to know of the concept of evaporation and its factors.</li> <li>➤ They would get knowledge regarding the reason why evaporation causes cooling.</li> <li>➤ They would acquire knowledge of the differentiation between boiling and evaporation.</li> </ul>	

	regarding the difference between evaporation and boiling.									
2. Evaporation and the factors affecting its rate.	<ul style="list-style-type: none"> <li>• Evaporation is defined as the change of the state of matter the liquid to vapour state at any temperature below its boiling point.</li> <li>• The Factors affecting the rate of Evaporation are: - <ul style="list-style-type: none"> <li>• Temperature</li> <li>• Humidity</li> <li>• Wind Speed</li> <li>• Surface Area</li> <li>• Nature of the Liquid</li> </ul> </li> </ul>									
3-Evaporation causes cooling.	<ul style="list-style-type: none"> <li>➤ Evaporation is a surface phenomenon in which a lot of heat.</li> <li>➤ Energy is required to overcome the attractive forces within the liquid molecules to convert into the vapour state.</li> <li>➤ The required energy is absorbed from the surrounding and thus causes cooling.</li> </ul>									
4.Difference between evaporation and Boiling.	<table border="1" style="margin-left: auto; margin-right: auto;"> <thead> <tr> <th>Evaporation</th> <th>Boiling</th> </tr> </thead> <tbody> <tr> <td>Surface phenomena</td> <td>Bulk phenomena</td> </tr> <tr> <td>At any temperature</td> <td>At 100-degree Celsius</td> </tr> <tr> <td>Causes cooling</td> <td>Do not causes cooling</td> </tr> </tbody> </table>	Evaporation	Boiling	Surface phenomena	Bulk phenomena	At any temperature	At 100-degree Celsius	Causes cooling	Do not causes cooling	
Evaporation	Boiling									
Surface phenomena	Bulk phenomena									
At any temperature	At 100-degree Celsius									
Causes cooling	Do not causes cooling									
5.Home Assignment	<ol style="list-style-type: none"> <li>1. Explain how does evaporation causes cooling.</li> <li>2. Explain the factors affecting the rate of evaporation.</li> </ol>									





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