

## DCP FOR CHAPTER-2

### PHYSICAL QUANTITIES AND MEASUREMENT

Number of periods	Sub-Topics
1	Density, density of a regular solid
2	Determination of density of an irregular solid, Density of a liquid, density bottle.
3	Density of liquid using density bottle, Relative density,
4	Floating and sinking, Principle of floatation, application of floatation.
5	Numerical on density.
6	Summarization of the chapter, Exercise questions.

<b>Class</b>	VIII	<b>Subject</b>	PHYSICS
<b>Prd</b>	1	<b>Chapter-2</b>	Physical Quantities and measurement
<b>Sub-Concepts</b>	Density, density of a regular solid		
<b>Teaching Aid To be used</b>	Smart Class, PowerPoint presentation.		
<b>Learning Outcome</b>	<p>Students will be able to</p> <ul style="list-style-type: none"> <li>➤ Measure density of a regular solid.</li> <li>➤ Define density.</li> <li>➤ State the unit of density.</li> </ul>		
<b>Sl. No</b>	<b>Step Wise (What to be done)</b>		
<b>1. Introduction</b>	<p><b>For Achievers</b></p> <ul style="list-style-type: none"> <li>➤ Give concept on mass and volume.</li> <li>➤ Interpret the word density.</li> </ul>	<p><b>For Average</b></p> <ul style="list-style-type: none"> <li>➤ Give concept on mass and volume.</li> <li>➤ Interpret the word density.</li> </ul>	
<b>Density</b>	<p>The teacher will explain the concept of density by showing a video.  <a href="https://youtu.be/QXoQbWoliRE">https://youtu.be/QXoQbWoliRE</a></p> <ul style="list-style-type: none"> <li>➤ State the formula for density.</li> <li>➤ State the SI unit of density.</li> <li>➤ Relationship between SI and CGS unit of density.</li> </ul>		
<b>Density of a regular solid</b>	<ul style="list-style-type: none"> <li>➤ The teacher will explain how to measure mass by using a beam balance.</li> <li>➤ Will explain how to measure volume</li> <li>➤ Volume of a cube</li> <li>➤ Volume of a sphere</li> <li>➤ Volume of a cylinder</li> </ul>		
<b>Home Assignment</b>	Exercise:B-1,2,3.		

<b>8. Common Errors</b>	Mass is measured by beam balance and weight is measured by spring balance
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<b>Class</b>	VIII	<b>Subject</b>	PHYSICS
<b>Prd</b>	2	<b>Chapter-2</b>	Physical Quantities and measurement
<b>Sub-Concepts</b>	Determination of density of an irregular solid, Density of a liquid, density bottle.		
<b>Teaching Aid To be used</b>	Smart Class, PowerPoint presentation.		
<b>Learning Outcome</b>	<p>Students will be able to</p> <ul style="list-style-type: none"> <li>➤ Determine density of an irregular solid.</li> <li>➤ Determine density of a liquid.</li> <li>➤ Determine density of a bottle.</li> </ul>		
<b>Sl. No</b>	<b>Step Wise (What to be done)</b>		
<b>1. Introduction</b>	<p><b>For Achievers</b></p> <ul style="list-style-type: none"> <li>➤ Recapitulation of the previous topic by asking the following questions.</li> <li>➤ Define the term density.</li> <li>➤ Name the SI unit of density.</li> </ul>	<p><b>For Average</b></p> <ul style="list-style-type: none"> <li>➤ Recapitulation of the previous topic by asking the following questions.</li> <li>➤ Define the term density.</li> <li>➤ Name the SI unit of density.</li> </ul>	
Determination of density of an irregular solid,	<ul style="list-style-type: none"> <li>➤ Demonstrate how to find out mass of a solid.</li> <li>➤ Demonstrate how to find out volume by using displacement method.</li> <li>➤ Explain how to calculate density.</li> <li>➤ <a href="https://youtu.be/s5u5cmA9Dp0">https://youtu.be/s5u5cmA9Dp0</a></li> </ul>		
Density of a liquid	<ul style="list-style-type: none"> <li>➤ Find the mass of the liquid.</li> <li>➤ Find the volume of the liquid by using a measuring cylinder.</li> <li>➤ <math>D = m/v</math></li> </ul>		
Density bottle	<ul style="list-style-type: none"> <li>➤ Show a density bottle and explain about its uses.</li> </ul>		

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<b>Home Assignment</b>	Exercise:B-1,2,3,4
<b>8. Common Errors</b>	Measurement of volume of liquid.

<b>Class</b>	VIII	<b>Subject</b>	PHYSICS
<b>Prd</b>	3	<b>Chapter-2</b>	Physical Quantities and measurement
<b>Sub-Concepts</b>	Density of liquid using density bottle, Relative density,		
<b>Teaching Aid To be used</b>	Smart Class, PowerPoint presentation.		
<b>Learning Outcome</b>	<p>Students will be able to</p> <ul style="list-style-type: none"> <li>➤ Determine density of liquid using density bottle.</li> <li>➤ Define relative density.</li> <li>➤ Solve numerical problems on relative density.</li> </ul>		
<b>Sl. No</b>	<b>Step Wise (What to be done)</b>		
<b>1. Introduction</b>	<p><b>For Achievers</b></p> <ul style="list-style-type: none"> <li>➤ Recapitulate the previous topic by asking the following questions.</li> <li>➤ Arrange the following substances in order of their increasing density: Iron, Cork, Brass, Water, Mercury. <ul style="list-style-type: none"> <li>➤ What is density bottle?</li> </ul> </li> </ul>	<p><b>For Average</b></p> <ul style="list-style-type: none"> <li>➤ Recapitulate the previous topic by asking the following questions.</li> <li>➤ Arrange the following substances in order of their increasing density: Iron, Cork, Brass, Water, Mercury.</li> <li>➤ What is density bottle?</li> </ul>	
Density of liquid using density bottle	Explain how to find out density of liquid by using a density bottle.		
Relative density	<ul style="list-style-type: none"> <li>➤ The teacher will show a video on how to find out relative density.</li> <li>➤ <a href="https://youtu.be/G6XQnlIwtt8">https://youtu.be/G6XQnlIwtt8</a></li> <li>➤ Find the unit of relative density.</li> </ul>		
<b>Home Assignment</b>	Exercise:B-10,11,12		

<b>8. Common Errors</b>	Relative density is just a number. It has no unit.
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<b>Class</b>	VIII	<b>Subject</b>	PHYSICS
<b>Prd</b>	4	<b>Chapter-2</b>	Physical Quantities and measurement
<b>Sub-Concepts</b>	Floating and sinking, Principle of floatation, application of floatation.		
<b>Teaching Aid To be used</b>	Smart Class, PowerPoint presentation.		
<b>Learning Outcome</b>	<p>Students will be able to</p> <ul style="list-style-type: none"> <li>➤ Understand the concept of floating and sinking</li> <li>➤ Explain the principle of floatation.</li> <li>➤ Apply this principle of floatation in making sub marine.</li> </ul>		
<b>Sl. No</b>	<b>Step Wise (What to be done)</b>		
<b>1. Introduction</b>	<p><b>For Achievers</b></p> <ul style="list-style-type: none"> <li>➤ Recapitulation of previous topic by asking the following questions.</li> <li>➤ Define relative density.</li> <li>➤ What is the SI unit of relative density?</li> </ul>	<p><b>For Average</b></p> <ul style="list-style-type: none"> <li>➤ Recapitulation of previous topic by asking the following questions.</li> <li>➤ Define relative density.</li> <li>➤ What is the SI unit of relative density?</li> </ul>	
Floating and sinking, Principle of floatation	<p>The teacher will explain the principle of floatation by showing a video.</p> <p><a href="https://youtu.be/khc2wUBsFU4">https://youtu.be/khc2wUBsFU4</a></p> <p><a href="https://youtu.be/2dyCe1GPagE">https://youtu.be/2dyCe1GPagE</a></p>		
Application of floatation.	<ul style="list-style-type: none"> <li>➤ Explain the following examples:</li> <li>➤ Floatation of iron sheep</li> <li>➤ Floatation of a man</li> <li>➤ Floatation of ice on water.</li> </ul>		
<b>Home Assignment</b>	Exercise:B-18,19,20		



<b>Class</b>	VIII	<b>Subject</b>	PHYSICS																
<b>Prd</b>	5	<b>Chapter-2</b>	Physical Quantities and measurement																
<b>Sub-Concepts</b>	Numerical on density.																		
<b>Teaching Aid To be used</b>	Smart Class, PowerPoint presentation.																		
<b>Learning Outcome</b>	Students will be able to <ul style="list-style-type: none"> <li>➤ Apply the concept of density in day to day life situations.</li> <li>➤ Solve numerical problems based on density.</li> </ul>																		
<b>Sl. No</b>	<b>Step Wise (What to be done)</b>																		
<b>1. Introduction</b>	<b>For Achievers</b> <ul style="list-style-type: none"> <li>➤ Recapitulate the previous topic by asking the following questions:</li> <li>➤ Define density.</li> <li>➤ What is the SI unit of density?</li> <li>➤ What is the other name of density bottle?</li> </ul>	<b>For Average</b> <ul style="list-style-type: none"> <li>➤ Recapitulate the previous topic by asking the following questions:</li> <li>➤ Define density.</li> <li>➤ What is the SI unit of density?</li> <li>➤ What is the other name of density bottle?</li> </ul>																	
<b>Numerical problems on density</b>	The table below shows the density of some solids and liquids. For each solids list the name of liquids in which that solid will Float Sink																		
	<table border="1"> <thead> <tr> <th>Solid</th> <th>Density in kg m<sup>-3</sup></th> <th>Liquid</th> <th>Density in kg m<sup>-3</sup></th> </tr> </thead> <tbody> <tr> <td>Iron</td> <td>7800</td> <td>Mercury</td> <td>13600</td> </tr> <tr> <td>Wood</td> <td>700</td> <td>Water</td> <td>1000</td> </tr> <tr> <td>Cork</td> <td>250</td> <td>Glycerine</td> <td>1260</td> </tr> </tbody> </table>	Solid	Density in kg m <sup>-3</sup>	Liquid	Density in kg m <sup>-3</sup>	Iron	7800	Mercury	13600	Wood	700	Water	1000	Cork	250	Glycerine	1260		
Solid	Density in kg m <sup>-3</sup>	Liquid	Density in kg m <sup>-3</sup>																
Iron	7800	Mercury	13600																
Wood	700	Water	1000																
Cork	250	Glycerine	1260																
	<ul style="list-style-type: none"> <li>➤ A block of silver displaces 200 mL of water in a measuring cylinder. If the density of silver is 10300 kg m<sup>-3</sup>, find the</li> </ul>																		

	mass of the block.
	➤ A block of glass is 30 cm long, 25 cm wide, and has a thickness of 2 cm. Find its density if its mass is 7.5 kg.
<b>Home Assignment</b>	Exercise: C-6,7,8
<b>8. Common Errors</b>	Putting the formula in different cases and solve the numerical.

<b>Class</b>	VIII	<b>Subject</b>	PHYSICS
<b>Prd</b>	6	<b>Chapter-2</b>	Physical Quantities and measurement
<b>Sub-Concepts</b>	Summarization of the chapter, Exercise questions.		
<b>Teaching Aid To be used</b>	Smart Class, PowerPoint presentation.		
<b>Learning Outcome</b>	<p>Students will be able to</p> <ul style="list-style-type: none"> <li>➤ Perform simple calculations to find density.</li> <li>➤ Measure mass and volume of a variety of solids and liquids and hence determine their densities.</li> <li>➤ Investigate floatation for a variety of solids and liquids in water and other liquids, and relate the results to density.</li> </ul>		
<b>Sl. No</b>	<b>Step Wise (What to be done)</b>		
<b>1. Introduction</b>	<p><b>For Achievers</b></p> <ul style="list-style-type: none"> <li>➤ Recapitulate the previous topic by asking the following questions.</li> <li>➤ State the law of floatation.</li> <li>➤ A cork piece floats on the surface of water but iron nail sinks. Why?</li> <li>➤ Distinguish between density and relative density.</li> </ul>	<p><b>For Average</b></p> <ul style="list-style-type: none"> <li>➤ Recapitulate the previous topic by asking the following questions.</li> <li>➤ State the law of floatation.</li> <li>➤ A cork piece floats on the surface of water but iron nail sinks. Why?</li> <li>➤ Distinguish between density and relative density.</li> </ul>	
<b>Exercise questions discussion</b>	<ol style="list-style-type: none"> <li>1. True/False type questions are to be discussed.</li> <li>2. Fill in the blanks type questions are to be discussed.</li> <li>3. Match the following type questions are to be discussed.</li> <li>4. MCQ are to be discussed.</li> </ol>		
<b>Home Assignment</b>	Exercise: A: 1,2,3,4		
<b>8. Common Errors</b>	Solving numerical problems.		

