

**WORKSHEET-4**  
**CHEMISTRY**  
**CLASS-VII**  
**ATOMS, MOLECULES AND RADICALS**  
**FM-30**

**Sub-Topic: - Characteristics of Atoms, Atomicity and Radicals**

**LEVEL-1 (1 Mark Each)**

**(I) Fill up the Gaps: (1X6=6)**

- 1) Atoms are\_\_\_\_\_
- 2) Metals have \_\_\_\_\_valency.
- 3) The number of atoms present in a molecule of an element is called as \_\_\_\_\_.
- 4) The combining capacity of an element is known as\_\_\_\_\_
- 5) \_\_\_\_\_is the element with valency 4
- 6) The valency of iron in  $\text{Fe}_2\text{O}_3$  is \_\_\_\_\_

**II) Define the following: - (1 Mark Each) (1X4=4)**

- 1) Atoms
- 2) Molecule
- 3) Valency
- 4) Radicals

**III) Write the symbols and valency's of the following radicals: (1X3=3)**

- 1) Magnesium ion
- 2) Carbonate
- 3) Ammonium

**IV) Name the Following radicals(1X5=5)**

- 1)  $\text{SO}_4^{2-}$
- 2)  $\text{OH}^-$
- 3)  $\text{HCO}_3^-$
- 4)  $\text{Cr}_2\text{O}_7^{2-}$
- 5)  $\text{PO}_4^{3-}$

**Sub-Topic-Periodic Table and Writing the chemical formula of a compound.**

**LEVEL-1 (1 Mark Each)**

- I) Write the names of the elements present in the following compounds. (1X4=4)
  - a) Common salt
  - b) Sulphuric acid
  - c) Sodium hydroxide
  - d) Ammonia
- II) Write the names of the following compounds. (1X3=3)
  - a)  $(\text{NH}_4)_2\text{SO}_4$
  - b)  $\text{FeS}$
  - c)  $\text{NH}_4\text{OH}$
- III) Write the molecular formula of the following compounds. (1X5=5)
  - a) Sodium sulphide
  - b) Magnesium oxide
  - c) Calcium hydroxide
  - d) Hydrogen chloride
  - e) Sulphuric acid

