

(1) Write true or false for each statement

1.  $1 \text{ g cm}^3 = 1000 \text{ kg m}^{-3}$

Ans → True

2. The density of water is maximum at  $4^\circ \text{C}$

Ans

3. The speed  $5 \text{ ms}^{-1}$  is less than  $25 \text{ km h}^{-1}$

Ans → True

4. The S.I. unit of speed is  $\text{ms}^{-1}$

Ans True

2. ~~Fill~~

2. Fill in the blanks.

I. The S.I. unit of density is  $\text{kg m}^{-3}$

II.  $1 \text{ g cm}^{-3} = \frac{1000}{1000} \text{ kg m}^{-3}$

III.  $36 \text{ km h}^{-1} = 10 \text{ ms}^{-1}$

IV. Distance travelled  $d = \text{_____} \times \text{time } t$

### 3. Short Answer Type Question

1. Define the term ~~mass~~ measurement?

Ans → The process of associating numbers with physical quantities and phenomena.

2. Define the term length. What is its S.I. unit?

Ans → The length is ~~the~~ the distance between two points. Its S.I. unit is metre.

3. Define the term mass. What is its S.I. unit?

The quantity of matter contained in the body is called mass. The S.I. unit of mass is Kilogram.

4. How will you determine the volume of a cuboid? Write the ~~the~~ formulae you will use.

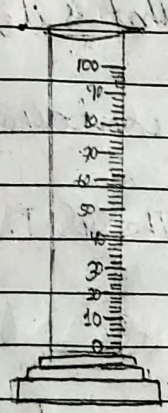
Ans → Volume of a cuboid = length  $\times$  breadth  $\times$  height.

5. Name two devices which are used to measure the volume of an object.  
Draw their neat diagrams.

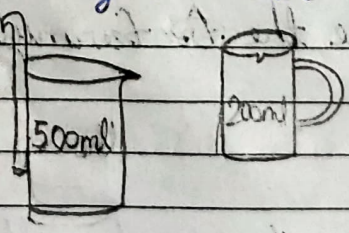
Ans Two devices that are used to measure the volume of an object are:

- (i) Measuring cylinder and
- (ii) Measuring beaker.

Measuring cylinder:-



Measuring beaker:-



(6) Find out the relation between  $m^3$  and  $cm^3$ ?

Ans: Cubic meter is equal to 1000000 cubic centimeters. To convert cubic meters to cubic centimeters, multiply the cubic meter value by 1000000.

7. The density of brass is  $8.4 \text{ g cm}^{-3}$ . What is the formula you will use. What do you

mean by the statement?

Ans: This statement means one cubic centimetre volume of brass has mass of  $8.4 \text{ g}$ .

8. Arrange the following substances in order of their increasing density: (a) iron (b) cork (c) brass (d) water (e) mercury.

Ans: The increasing order of the density is:

cork, water, Iron, Brass, Mercury.

9. How does the density of water changes when : (a) it is heated from  $0^{\circ}\text{C}$  to  $4^{\circ}\text{C}$ , (b) it is heated from  $4^{\circ}\text{C}$  to  $10^{\circ}\text{C}$ ?

Ans. When water is heated from  $0^{\circ}\text{C}$ , its volume decreases because its ~~density~~ density increases and you can see this effect upto  $4^{\circ}\text{C}$ .

Afterwards as the density decreases the volume increases.

10. A car travels with a speed  $12\text{ms}^{-1}$  while a scooter travels with a speed  $36\text{kmh}^{-1}$ .

Which of the two travels faster?

(Speed of car)  $= 12\text{ms}^{-1}$ . Speed of scooter  $= 36\text{kmh}^{-1}$  (here  $1\text{km} = 1000\text{m}$  and  $1\text{hr} = 3600$

See Speed of scooter  $= 36 \times 1000 \times 3600 = 10\text{ms}^{-1}$

$\therefore$  Speed of car is more. Car travels faster than scooter

1. Find out the relation between m/s and km/h?

Ans 1 (Kilometer/hour) = 1000 (meters) /

3600 (second) can also be express as

4 (Kilometer/hour) =  $\frac{5}{18}$  (meters/second)

(meters/second), which is its simplified

form. To convert km/h to m/s, directly

multiply the given value of speed by the

fraction  $\frac{5}{18}$

2. What is the S.I. unit of volume and density?

State other two smaller unit of volume?

Ans 1 The S.I. unit of volume is cubic meter. Two

smaller units of volume are: cubic centimetre

and cubic decimetre.

The S.I. unit of density is  $\frac{\text{Kg}}{\text{m}^3}$

Two smaller unit of.

14. A rectangle

13. Convert  $72 \text{ km/h}$  to  $\text{m/s}$

Ans  $72 \text{ km/hr} = 72 \times 1000 / 3600 = 20 \text{ m/s}$

(14) A rectangular park is of length  $30 \text{ m}$  and breadth  $5 \text{ km}$ . Find the area of the rectangular park?

Ans Given

Length of the park  $= 50 \text{ m}$

Breadth of the park  $= 30 \text{ m}$

Area of the park  $= 50 \times 30$

$= 1500 \text{ m}$

15. By what apparatus can we measure volume.

Name any two.

Ans ~~Measuring~~ Measuring jar, scales are

used to measure the volume of an object.

#### 4. Long Answer Type Questions

I Describe the method in steps to find the area of an irregular lamina using a graph paper.

Ans (1) place the lamina over a graph paper.

(2) Draw its boundary line on graph paper with a pencil.

3. Remove the lamina and count, note the number of complete squares as well as the number of squares more than half within the boundary line.

4. The area of the lamina is equal to the sum of the area of complete squares and the area of squares more than the half.

5. Let  $n$  be the total number of complete squares and more than half or half square



6. Area of lamina will be  $n \times$  as area of one big square =  $1 \text{ cm}^2$

II. How can you determine the volume of an irregular solid (say a piece of brass)?

Describe in steps with neat diagrams

Ans To measure the volume of a piece of stone:

1. Take a piece of <sup>brass</sup> stone, a measuring cylinder, fine thread of sufficient length and some water.

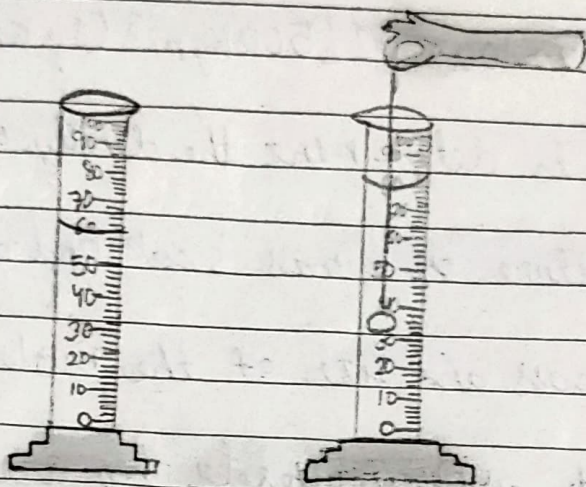
2. Fill it partially with water.

3. Note the reading of the water level very carefully.

4. Now tie the piece of brass with a thread and dip it completely into water. We see that the level of water rises.

5. Note the reading of the new water level

6. The difference in the initial and final water level readings give us the volume of stone.



III The mass of a lead piece is 115g. When it is immersed into a measuring cylinder, the water level rises from 20ml mark to 30ml mark.

Find: (i) The volume of the lead piece, (ii) The density of the lead in  $\text{Kg m}^{-3}$ .

Ans: (i) Given,  $M = 115\text{g}$

$$V_1 = 20\text{ml}, V_2 = 30\text{ml}$$

Volume of lead piece  $V = V_2 - V_1$

$$= 30\text{ml} - 20\text{ml}$$

$$= 10\text{ml or } 10\text{cm}^3$$

(ii) Density of lead piece  $= \frac{M}{V}$

$$= \frac{115}{10 \text{ cm}^3} = 11.5 \text{ g cm}^{-3} \quad (1 \text{ g cm}^3 = 1000 \text{ kg}^{-3})$$

$$= 11.5 \times 1000 = 11500 \text{ kg m}^{-3} \quad (1 \text{ g cm}^{-3} = 1000 \text{ kg}^{-3})$$

IV. How to determine the density of a liquid?

Ans: The mixture on a mass scale and read its mass.

Ans: The mass and size of the molecules in a liquid and how closely they are packed together determine the density of the

liquid. Just like a solid, the density of

a liquid equals the mass of the liquid

(divided by its volume);  $D = \frac{m}{V}$ . The density

of water is 1 gram per cubic

centimeter.