

C. Numericals

① The density of air is $1.28 \text{ g litre}^{-1}$. Express it in

a) g cm^{-3}

sol: Density of air = 1.28 g/l

$$\text{In } \text{g cm}^{-3} = \frac{1.28}{1000} = 0.00128 \text{ g cm}^{-3}$$

b) kg m^{-3}

$$= 0.00128 \times 1000 = 1.28 \text{ kg/m}^3.$$

② The dimensions of a hall are $10 \times 7 \times 5 \text{ m}$. If density of air is 1.1 kg m^{-3} . Find mass of air in the hall

sol: Dimensions of hall = $10 \times 7 \times 5 \text{ m}$	$M = V \times D$	
Volume of hall = 350 m^3		$= 350 \times 1.1$
Density of air = 1.1 kg m^{-3}		$= 385 \text{ kg}$

③ The density of aluminium is 2.7 g cm^{-3} . Express it in kg m^{-3} .

sol: density of Al = 2.7 g/cm^3

$$\text{In } \text{kg/m}^3 = \frac{2.7 \times 1000}{10} = 2700 \text{ kg m}^{-3}$$

④ The density of alcohol is 600 kg/m^3 . Express it in g cm^{-3} .

sol: Density of alcohol = 600 kg/m^3

$$\text{In } \text{g/cm}^3 = \frac{600}{1000} = 0.60 \text{ g/cm}^3$$

⑤ A ~~piece~~ piece of zinc of mass 438.6 g has a volume of 86 cm^3 . Calculate the density of zinc.

sol: Mass of zinc (m) = 438.6 g

$$V = 86 \text{ cm}^3$$

$$D = \frac{m}{V} = \frac{438.6}{86} = 5.1 \text{ g/cm}^3$$

⑥ A piece of wood of mass 150 g has a volume of 200 cm^3 . Find density of wood in CGS unit & SI unit.

sol: Mass of wood = 150 g

$$\text{Volm of wood} = 200 \text{ cm}^3$$

$$\text{Density} = \frac{m}{V} = \frac{150}{200} = 0.75 \text{ g/cm}^3$$

$$\text{In SI system} = 0.75 \times 1000 = 750 \text{ kg/m}^3$$