

(1) * Weak acids :- These acids do not completely dissociate its ions in aqueous solution.
Ex- carbonic acid, acetic acid.

* Strong acids :- These acids get dissociate into ^{its ions} completely in aqueous solution.
Ex- hydrochloric acid, sulphuric acid.

(2) Amphoteric oxides :- These are the compounds which react with both acid and bases.

