

Determination of density of an irregular solid, Density of a liquid, density based on

Home assignment

3) The density of alcohol is 600 kg m^{-3} . Express it in g cm^{-3} .

Ans: The density of the alcohol is g cm^{-3}

$$\begin{aligned} 1 \text{ kg m}^{-3} &= 10^{-3} \text{ g cm}^{-3} \\ 600 \text{ kg m}^{-3} &= 10^{-3} \text{ g cm}^{-3} \end{aligned}$$

$$\begin{aligned} &= 600 \times 10^{-3} \text{ g cm}^{-3} \\ &= 6000 \text{ g cm}^{-3} \end{aligned}$$

2) A piece of wood of mass 150 g has a volume of 200 cm^3 . Find the density of wood in a) CGS unit b) S.I unit

Answers: $D = \frac{M}{V}$

$$D = \frac{150}{200} = \frac{3}{4} = 0.75 \text{ g cm}^{-3}$$

In SI unit water = $1 \text{ g cm}^{-3} = 1000 \text{ kg m}^{-3}$

$$0.75 \text{ g cm}^{-3} = 0.75 \times 1000 \\ = 750 \text{ kg m}^{-3}$$

3) Calculate the density of solid from the following data:

a) Mass of solid (M) = 72g

b) Initial volume of water in measuring cylinder = 24ml

c) Final volume of water when solid is completely immersed in water = 48ml

∴ Final volume of the completely immersed water = 48ml

Initial volume = 24ml

Volume of solid = $V_2 - V_1$

$$= 48 - 24$$

$$18 \text{ ml} = 18 \text{ cm}^3$$

$$D = \frac{M}{V}$$

$$0.2 \frac{\text{g}}{\text{cm}^3} \times 18 \text{ cm}^3 = 3.6 \text{ g}$$

4) How does the density of a liquid (or gas) vary with temperature?

Answer: The density of a liquid or gas varies due to temperature because increase in the temperature makes increase in the volume due to which there is ~~less~~ more density. When the temperature decreases then the volume also decreases for which the density also decreases.

5) What is a density bottle? How is it used to find the density of a liquid?

Answer: The density bottles are normally used as ~~for~~ for measuring the viscosity of the liquid.