

Q1(i) Mass of silver = 1000 gm = 1 kg
 volume of silver = $10 \text{ cm}^3 = \frac{10}{1000000} = 10^{-5}$

Density = $\frac{\text{Mass}}{\text{volume}} = \frac{1}{10^{-5}} = 10^5 \text{ kg/m}^3$

(ii) Relative Density = $\frac{\text{Density of silver}}{\text{Density of water}}$

Density of water = $\frac{1 \text{ kg}}{1 \text{ m}^3} = \frac{10^3}{1} = 10^3$

Q2(i) Mass of wood = 150 g
 volume of wood = 200 cm^3
 Density = $\frac{M}{V} = D = \frac{150}{200} = D = 0.75 \text{ g/cm}^3$

(ii) In SI system = $0.75 \times 1000 = 750 \text{ kg/m}^3$

Q3) As the temperature increases, volumes of most of liquids also increases and when the volume increases density decreases. Similarly, when temperature decreases, the volume of most liquids decreases increases the density. However, water shows an anomalous behaviour. water has maximum volume at 4°C.

celcius and maximum density at 4 degree celcius. But when water is cooled down further its volume starts increasing and hence, the density of water decreases when cooled further below than 4 degree celcius.

Hence, the density of water is maximum at 4 degree celcius at $\frac{1}{1}$ or $1000/\text{kg}/\text{m}^{-3}$
 $\frac{1}{\text{cm}^{-3}}$

Q4) Relative Density is the ratio of mass of the substance to the mass of an equal volume of at 4°C. It is a ratio and thus has no units.

Q5) The Density of an object determines whether it will float or sink in another substance. An object will float if it is less dense than the liquid it is placed in. An object will sink if it is more dense than the liquid it is placed in.

Q6) The principle of floatation states that when an object floats on a liquid the buoyant force that acts on the object is equal to the weight of the object.

Q7(a) the buoyant force is the same in each case as the weight of the body is the same in each case and buoyant force is equal to the weight of liquid displaced by the immersed part of the body which balances the weight of the body.

(b) The liquid A has the least density as the body immerses the maximum.

(c) liquid C has the highest density as the body immerses the least.