

## ASSIGNMENT - 1

- ① → The density of a substance is its mass per volume.
- ② → S.I unit of density is  $\text{m}^{-3}$  in CGS system unit of mass is g and unit of volume is  $\text{cm}^3$ , so CGS unit of density is  $\text{g cm}^{-3}$  (gram per cubic centimetre)
- ③ → The statement means one cubic centimetre volume of brass has mass of 8.4g.
- ④ → The order is :- cork, water, iron, brass, and mercury.
- ⑤ → Most of the liquids increase in volume with increase in temperature, but water shows odd behaviour. water has its volume and density both at  $4^\circ\text{C}$ . when the volume increases density decreases and when the density increases volume decreases.
- ⑥ → when a given quantity of liquid is

heated, the mass doesn't change. The volume changes and increases with rise in temperature. The density changes & decreases.

- ⑦ → To find the density of the coin material, we need to find its mass by beam balance and its volume by measuring cylinder.

### EXPERIMENT -

Let the initial volume of water be =  $V_1 = 42 \text{ ml}$ .

Let the final volume of water be when the coin is added in the cylinder =  $V_2 = 52 \text{ ml}$ .

So, the volume of the coin is =  $V_2 - V_1$ ,

$$= 52 - 42 = 10 \text{ ml.}$$

Density of the coin = 10

$$= D = \frac{M}{V} = \frac{50}{52 - 42} = \frac{50}{10} = 5 \text{ g cm}^{-3}$$

⑧ → To find the density of a liquid =  
 $D = M/V$ .

We need to find the volume of liquid.  
 For ex - milk.

The bottle is filled with water and at last the bottle is weighted with the liquid and the density is to be determined.

Experiment:-

To find the mass of milk =  
 $M_1 \text{ g} = 70 \text{ g}$  (wt)

Fill the beaker half with milk and  
 measure =  $M_2 \text{ g} = 116 \text{ g}$  (elt)

Now, to find the volume of milk into measuring cylinder & note the volume  $V$   
 $= 40$  (elt)

$$= \text{Density of milk} = D = \frac{M}{V} = \frac{M_2 - M_1}{V}$$

$$= \frac{(116 - 70)}{40} = \frac{46}{40} = \frac{46}{4} = 1.15 \text{ g cm}^{-3}$$

⑨ → Density bottles are used to determine the density of the liquids.

It makes use of the bottles weight and