

## Homework

1) The density of alcohol is  $600 \text{ g/kg} / \text{m}^{-3}$ . Express it in  $\text{g/cm}^{-3}$ .

Ans Density of alcohol is  $= 600 \text{ kg/m}^{-3}$

$$\text{In g/cm}^{-3} = \frac{600}{1000} = 0.60 \text{ g/cm}^{-3}$$

2) A piece of wood of mass 150 g has a volume of  $200 \text{ cm}^3$ . Find the density of wood in  
(a) C.G.S. unit, (b) S.I. units.

Ans a) Mass of wood = 150 g

Volume of wood =  $200 \text{ cm}^3$

Density =  $m/v$

$$D = 150 / 200$$

$$\text{Density} = 0.75 \text{ g/cm}^3$$

b) In S.I. system  $= 0.75 \times 1000 = 750 \text{ kg/m}^3$

3) Calculate the density of solid from the following data:

a) Mass of solid (M) = 72 g

b) Initial volume of water in measuring cylinder = 24 mL

c) Final volume of water when solid is completely immersed in water 42 ml.

Ans Mass of solid ( $M$ ) = 72 g

Initial volume of water  $V_1$  = 24 mL

Final volume of water  $V_2$  = 42 mL

$$\begin{aligned} \text{Volume of solid } (V) &= V_2 - V_1 \\ &= 42 - 24 \\ &= 18 \text{ cm}^3 \end{aligned}$$

$$\text{Density of solid } (D) = \frac{M}{V}$$

$$= \frac{72}{18}$$

$$= 4 \text{ g/cm}^3$$

4) How does the density of a liquid (or gas) vary with temperature?

Ans As the temperature increases, volume of most of the liquids also increases and when the volume increases density decreases. Similarly when temperature decreased, the volume of most liquids decreases which increase the density.