

Daily Homework - 16/07

$$1. \quad 600 \text{ kg/m}^3 = \frac{600 \text{ kg}}{1 \text{ m}^3} = \frac{600 \times 1000 \text{ g}}{1000 \times 1000 \times 1000 \text{ cm}^3}$$
$$= 0.6 \text{ g/cm}^3.$$

2. Mass of wood = 150 g
Volume of wood = 250 cm³

$$2) \quad \text{Density} = \frac{m}{V} = \frac{150 \text{ g}}{250 \text{ cm}^3} = \frac{3}{5} \text{ g/cm}^3 \text{ (CGS unit)}$$

$$\frac{3}{5} \text{ g/cm}^3 = \frac{3}{5} \times \frac{250}{1000} \text{ kg/cm}^3$$
$$= 750 \text{ kg/cm}^3 \text{ (SI unit)}.$$

3. Given,

→ mass = 72 g

→ Initial Volume of water = 24 ml.

→ final Volume of water = 42 ml.

$$\therefore \text{Vol}^{\text{me}} \text{ of solid} = 42 \text{ ml} - 24 \text{ ml}$$
$$= 18 \text{ ml}.$$

$$= 18 \text{ cm}^3$$

$$2) \quad \text{Density} = \frac{M}{V} = \frac{72 \text{ g}}{18 \text{ cm}^3} = 4 \text{ g/cm}^3.$$

g,

A density bottle is a glass instrument used to find the density of any liquid. The bottle has a stopper and a glass body. When liquid is inserted inside it, it stores a fixed amount and rest of it drains out through the hole in its stopper. It is generally found of volume 50 ml / 25 ml.