

CHAPTER - 4

ELEMENTS, COMPOUNDS, SYMBOLS & FORMULAE

Date _____

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Objective type Questions :-

1. Fill in the blanks :-

- a) Atomicity refers to the number of atoms in the molecule of an element.
- b) The most abundant element in the earth's crust is Oxygen.
- c) A metal which is a liquid at room temperature is mercury.
- d) The most abundant element in the atmosphere is nitrogen.
- e) A metal which is a poor conductor of electricity is tungsten.
- f) A diatomic gaseous element is oxygen.
- g) A liquid non-metal is bromine.

2. Match the Columns :-

Column A

- a) Metals
- b) Molecules
- c) Non metals
- d) Noble gases

Column B

- i) Non-reactive (d)
- ii) Brittle (c)
- iii) Lustrous (a)
- iv) Smallest unit of Compound - (b)

Answers : a (iii) b (iv) c (ii) d (i)

3. State True or false :

- a) A Compound is made up of just one kind of atom . False (made up of 2 or more elements in a fixed proportion by mass)
- b) Metals reflect light and are good conductor of electricity . True
- c) Metals can be polished . True
- d) Elements are made up of compounds . False (made of atoms)
- e) All elements are artificially prepared . False (made up of a limited number of basic substances)
- f) Molecules can exist independently . True

- g) Molecules combine to form atoms. False
(form molecules)
- h) Noble gases are highly reactive. False
(non-reactive)
- i) Ozone is a triatomic molecule. True

Multiple choice questions:-

1. All pure substances have
 - a) the same physical state
 - b) the same colour
 - c) the same composition
 - d) a definite set of properties
2. Sugar is a compound which consists of the elements :
 - a) carbon and hydrogen
 - b) hydrogen and oxygen
 - c) carbon, hydrogen and oxygen
 - d) hydrogen, carbon and sulphur.
3. Atoms of different kinds combine to form molecules of
 - a) an element
 - b) a compound
 - c) a mixture
 - d) all of the above

4. Sulphur and Carbon are

- a) metals
- b) non-metals
- c) metalloids
- d) noble gases

5. Gold is used to make jewellery because

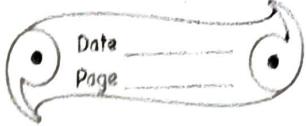
- a) it is dull
- b) lustrous and attractive
- c) highly reactive
- d) very cheap

6. The most abundant elements in the universe are

- a) neon and argon
- b) hydrogen and helium
- c) aluminium and copper
- d) oxygen and nitrogen

7. The compound used as common salt is

- a) sodium chloride
- b) calcium chloride
- c) sodium oxide
- d) hydrogen chloride



8) Brass and bronze are

- a) elements
- c) compounds

b) mixtures

d) All of the above.

9) Sand is a compound of

- a) silicon and nitrogen
- b) silicon and oxygen
- c) oxygen and Sulphur
- d) none of the above.

10) From the list given below Select the correct substance which is most suitable to the statement given :

[oxygen, diamond, zinc, graphite, gold]

- a) A metal which is brittle. Zinc
- b) A non-metal which is a good conductor of electricity. graphite
- c) The hardest naturally occurring substance. Diamond.
- d) The most ductile metal. Gold
- e) A gaseous non-metal. oxygen.

Exercise - II

Q7. Write the molecular formulae of compounds Calcium oxide, hydrogen sulphide, carbon monoxide and lead sulphite.

Ans. Calcium oxide - CaO

hydrogen sulphide - H_2S

carbon monoxide - CO

lead sulphite - PbS

Q8) Give two examples each of compounds existing in the following states:

(a) Solid : NaCl , SiO_2

(b) Liquid : H_2O , HCl

(c) Gaseous : CO_2

a) Write formulas of :

