

H/W
9/9/21

PURE SUBSTANCES AND MIXTURES ; SEPARATION OF MIXTURE

Q1) Differentiate between pure substances and mixtures?

Ans- Pure Substances

- * Pure Substances can be categorised as element and compound;
- * It cannot be broken down or separated into new products.
- * Constant physical and chemical properties
- * Pure Substances are made up of a single element.
- * Ex - iron, steel, water

Mixtures

- * Mixtures can be categorised as only homogenous and heterogenous.
- * It can be separated using different separation methods.
- * Mixtures have varying physical and chemical properties
- * A mixture is a combination of two substances or elements.
- Ex - wood

Q2) Give 3 examples each of homogeneous and heterogeneous mixture?

Ans- Heterogeneous - orange juice with pulp, pizza, sandwich.

Homogeneous — air, saline solution, alloys

Q3) Explain the principle of the process of winnowing for separation of solid-solid mixture?

Ans - Winnowing is the process to separate heavier particles from the lighter particles of a mixture. This method is used by the farmers to separate lighter seeds from heavier seeds of grains.

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