

## Home assignment

1 The density of alcohol is  $600 \text{ kg/m}^{-3}$ .  
Express it in  $\text{g cm}^{-3}$ .

$$\Rightarrow \text{Density of alcohol is} = 600 \text{ kg/m}^{-3} \text{ in } \text{g/cm}^{-3} = 600/1000 = 0.60 \text{ g/cm}^{-3}$$

2 A piece of wood of mass  $150 \text{ g}$  has a volume of  $200 \text{ cm}^3$ . Find the density of wood in (a) C.G.S unit, (b) S.I. unit.

$$\Rightarrow \text{Mass of wood (M)} = 150 \text{ g}$$

$$\text{Volume of wood (V)} = 200 \text{ cm}^3$$

$$\text{Density (D)} = ?$$

$$D = \frac{M}{V} = \frac{150}{200} = 0.75 \frac{\text{g}}{\text{cm}^3}$$

$$\text{In S.I. system} = 0.75 \times 1000$$

$$= 750 \text{ kg/m}^3$$

3. Mass of solid (M) = 72 g

Initial volume of water  $V_1 = 24 \text{ ml}$

Final volume of water  $V_2 = 42 \text{ ml}$

Volume of solid (V) =  $V_2 - V_1 = 42 - 24 = 18 \text{ cm}^3$

Density of solid (D) = ?

$$D = \frac{M}{V} = \frac{72}{18} = 4.0 \text{ g cm}^{-3}$$

4. Most of the liquid increase in volume with an increase in temperature, but water shows anomalous behaviour - water has a maximum volume at  $4^\circ\text{C}$  and maximum density at  $4^\circ\text{C}$ . Actually, when volume ~~de~~ increases density decreases.

5. Density bottle is a small glass bottle which has a glass stopper at its neck.

1. Measure the volume and mass of a liquid, and calculate  $D = \frac{m}{V}$