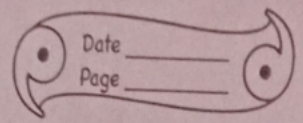


Exercise-1



1. Saturated solution is sand and water

2.

a) Pure substance - pure substances are either elements or compounds. They contain the same kind of atoms and molecules and have a definite set of physical and chemical properties.

b) Impure substance - A substance in which some other substances are also present in smaller or larger amounts is called a mixture or impure substance.

c) Alloy - A homogeneous (solid) mixture of two or many metals or a metal and a non-metal is called Alloy.

d) Solution - The homogeneous mixture of water and a substance soluble in it is called a solution.

e) Heterogeneous mixture - A mixture in which the components or constituents are not uniformly distributed throughout its volume and can be easily seen separately is called a heterogeneous mixture.

f) Homogeneous mixture - A mixture in which its constituents are uniformly

distributed throughout its volume and cannot be seen separately is called a homogeneous mixture.

PDF Created Using



Camera Scanner

Easily Scan documents & Generate PDF



<https://play.google.com/store/apps/details?id=photo.pdf maker>