

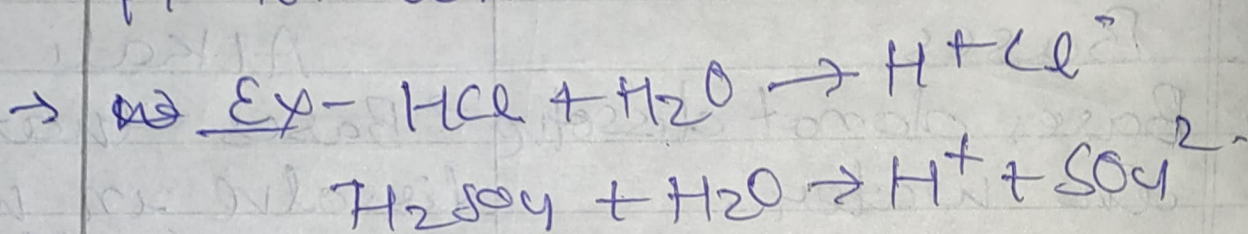
HW

03/07/21

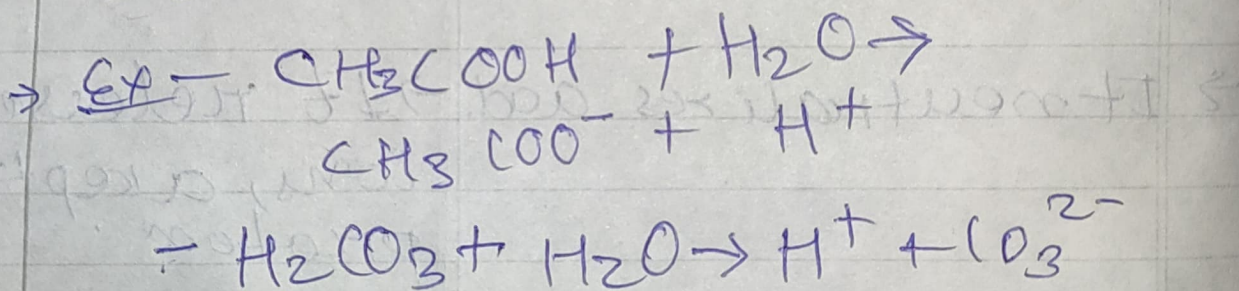
Home Assignment.

① What are weak acids and strong acids?

Ans - Strong acids is an acid that ionizes completely in aqueous solution. It always loses a H^+ ion when dissolved in water.



→ Weak acid is an acid that ionizes partially in a solution.



② What are amphoteric oxides? Explain with one such example.

Ans - Amphoteric oxides are the oxides that behave as both acidic and basic oxides.

