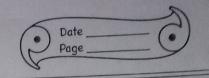
	Homeworck	Chapter-2 Page Page
1	- Is matter Arround Us Pune Page Give major 5 differences Between elements and compounds.	
	Element	Compound
	Composed of same type of atoms.	made up of by combin- ation of two or morce elements in definte mass mass by reatio volume.
	They are represented by Symbols	They are represented by chemical formulas
	It cannot be broken down into more simpler form. by Chemical methods.	the constituents cannot be separated by physical methods except by some chemical methods.
	mey exist in atomic 1 moleculare form.	They exist freely 9n naturel.
	The elements shows Same properties as of the atoms. Ex > H2, N2, D2	The constituting elements loose their properties in compounds. Ex > H20, Nacl, H2S04
1		



- 2> Explain why we say 14,0 as a compound whereas H2 and O2 as elements?
 - We say 2H,0 as a compound because it is made by the chemical combination of elements H2 and O2. in a definte mass by reatio volume.
- of H2 and O2.
 - It is represented by a chemical formula.
 - But we say H, and O2 as elements because it is composed of only one type of atoms. Like H2 is composed of only hydrogen atoms and O2 is composed of only oxygen atoms.
 - The elements H2 and O2 Shows, the same type of properties of the atoms of which they are composed of.
 - They are represented by symbols. Like by drogen element is represented by H2 whereas oxygen is represented by D2